





Journal of Computational and Applied Research in Mechanical Engineering Vol. 3, No. 1, pp. 1- 11, Autumn 2013 http://jcarme.srttu.edu

# Effect of welding parameters on pitting corrosion rate of pulsed current micro plasma arc welded AISI 304L sheets in 1N HCl

Kondapalli Siva Prasad<sup>a\*</sup>, Chalamalasetti Srinivasa Rao<sup>b</sup> and Damera Nageswara Rao<sup>c</sup>

<sup>a</sup>Department of Mechanical Engineering, Anil Neerukonda Institute of Technology & <sup>Sciences</sup>, Visakhapatnam, India <sup>b</sup>Department of Mechanical Engineering, AU College of Engineering, Andhra University, <sup>Visakhapatnam,</sup> India

<sup>c</sup>Centurion University of Technology & Management, Odisha, India

Article info:	Abstract
Received: 10/08/2012	Austenitic stainless steel sheets have gained wide acceptance in the fabrication
Accepted: 08/04/2013	of components, which require high temperature resistance and corrosion
Online: 11/09/2013	resistance such as metal bellows used in expansion joints in aircraft, aerospace
	and petroleum industries. In the case of single pass welding of thinner sections
	of this alloy, Pulsed Current Micro Plasma Arc Welding (PCMPAW) has been
Keywords:	found beneficial due to its advantages over the conventional continuous
AISI 304L,	current process. This paper highlighted development of empirical
Pulsed current,	mathematical equations using multiple regression analysis, correlating various
Micro plasma arc welding	process parameters to pitting corrosion rates in PCMPAW of AISI 304L
D'u	sheets in 1 Normal HCl. The experiments were conducted based on a five
Pitting corrosion.	factor, five level central composite rotatable design matrix. The model
	adequacy was checked by Analysis of Variance (ANOVA). The main effects
	and interaction effects of the welding process parameters on pitting corrosion
	rates of the welded joints were studied using surface and contour plots. From
	the contour plots, it was understood that peak current was the most influencing
	factor on the pitting corrosion rate. The optimum pitting corrosion rate was
	achieved at peak current of 6 Amperes, base current of 4 Amperes, pulse rate
	of 40 pulses/second and pulse width of 50 %.

# 1. Introduction

AISI 304L is an austenitic stainless steel with excellent strength and good ductility at high temperature. Its typical applications include aero-engine hot section components, miscellaneous hardware, tooling and liquid rocket components involving cryogenic temperature. AISI 304 L can be joined using a variety of welding methods including Gas Tungsten Arc Welding (GTAW), Plasma Arc Welding (PAW), Laser Beam Welding (LBW) and Electron Beam Welding (EBW). Among these methods, low current PAW (Micro PAW) has attracted particular attention and has been extensively used for the fabrication of metal bellows and diaphragms which require high strength and toughness. PAW is conveniently carried out using one of the two different current modes, namely a Continuous Current (CC) mode or a Pulsed Current (PC) mode.

Pulsed current MPAW involves cycling the welding current at the selected regular frequency. The maximum current is selected to give adequate penetration and bead contour while the minimum is set at a level sufficient for maintaining a stable arc [1, 2], which permits arc energy to be used effectively for fusing a spot of controlled dimensions in a short time, producing the weld as a series of overlapping nuggets. In pulsed current welding, the heat required for melting the base material is supplied only during the peak current pulses. allowing for the heat to dissipate into the base material leading to narrower Heat Affected Zone (HAZ). The advantages include improved bead contours, greater tolerance to heat sink variations, lower heat input requirements, reduced residual stresses and distortion. refinement of fusion zone microstructure and reduced width of HAZ. Based on the worked published as [3-8], four independent parameters that influence the process are peak current, back current, pulse rate and pulse width.

Neusa Alonso-Falleiros et al. [9] examined effect of surface finish of two AISI 304L (UNS S30403) stainless steels on the corrosion potential (Ecorr) in 3.5% NaCl aqueous solution. B. Tsaneva et al. [10] studied influence of temperature  $(0-80^{\circ}C)$ on corrosion electrochemical parameters of austenitic Cr/4Mn/5N and Cr/8Mn/2N stainless steels in 3.5% NaCl by cyclic potentiodynamic method. P. Fauvet et al. [11] analyzed various austenitic stainless steel types 304L, 316L and 310Nb and noticed that austentic stainless steels were largely used as structural materials for the equipment handling nitric acid media in reprocessing plants. D. J. Lee et al. [12] investigated effect of pitting corrosion behavior on welded joints of AISI 304L austenitic stainless steel by the flux-cored arc welding process. Effect of welding parameters (power input, weld geometry, welding speed and postweld heat treatment) on the corrosion behavior

of austenitic stainless steel in chloride medium was investigated by Ayo Samuel Afolabi [13]. Yunan Prawoto et al. [14] carried out a corrosion test to study performance of a duplex stainless steel alloy under several conditions using various pH and chloride concentrations at different temperatures. Girija Suresh et al. [15] conducted Electrochemical Noise (EN) monitoring of 304L stainless steel (SS) and sensitized 304 SS in 3N nitric acid and nuclear near-high level waste solution using a three nominally identical electrode configuration under open circuit conditions. Y. Ait Albrimi et al. [16] investigated electrochemical behavior of AISI 316 austenitic stainless steel in deaerated hydrochloric and sulphuric acid solutions using open-circuit potential, cvclic chronoamperometric voltammetric and techniques. Md. Asaduzzaman et al. [17] investigated the pitting corrosion behavior of the austenitic stainless steel in aqueous chloride solution using electrochemical technique. Moreover, M. Saadawy [18] studied effect of chloride ion addition on the corrosion of stainless steel 304 in Na<sub>2</sub>SO<sub>4</sub> solution under constant ionic strength conditions at 30°C using potential-time and potentiodynamic polarization techniques.

In this investigation, the experiments in the design of experimental concept were used for developing mathematical models to predict such variables. Many works have been reported in the past for predicting bead geometry, heataffected zone, bead volume, etc. using mathematical models for various welding processes [19-21]. Usually, the desired welding process parameters are determined based on the experience of skilled workers or from the data available in the handbook. This does not ensure formation of optimal or near-optimal weld pool geometry [22]. It has been proven by several researchers that efficient use of statistical design of experimental techniques and other optimization tools can impart scientific approach in a welding procedure [23, 24]. These techniques can be used to achieve optimal or near-optimal bead geometry from the selected process parameters.

Kim et Al. mentioned that optimization using regression modeling, neural network and

Taguchi methods could be effective only when the welding process was set near the optimal conditions or in a stable operating range [25]; but, near-optimal conditions could not be easily determined through full-factorial experiments when the number of experiments and levels of variables were increased. Also, the method of steepest ascent based upon derivatives could lead to incorrect direction of search due to nonlinear characteristics of the welding process. The main objective of the present work was to study the main and interaction effects of PCMPAW parameters on pitting corrosion rate in 1N HCl medium.

#### 2. Experimental procedure

Austenitic stainless steel (AISI 304L) sheets of 100 x 150 x 0.25 mm were welded autogenously with square butt joint without edge preparation. The chemical composition of SS304L stainless steel sheet is given in Table 1. High purity argon gas (99.99%) was used as a shielding gas and a trailing gas right after the welding to prevent absorption of oxygen and nitrogen from the atmosphere. Welding was carried out under the welding conditions presented in Table 2. From the literature, four important factors of pulsed current MPAW as presented in Table 3 were chosen. A large number of trail experiments were carried out using 0.25 mm thick AISI 304L sheets to find out the feasible working limits of pulsed current MPAW process parameters. Due to the wide range of factors, it was decided to use a four factor, five level, rotatable central composite design matrix to perform a number of experiments for the purpose of this investigation. Table 4 indicates 31 sets of coded conditions used for forming the design matrix. The first sixteen experimental conditions (rows) were formed for main effects. The next eight experimental conditions were called corner points and the last seven ones were known as

center points. The method of designing such a matrix has been mentioned in [26, 27]. For the convenience of recording and processing the experimental data, upper and lower levels of the factors were coded as +2 and -2, respectively, and the coded values of any intermediate levels was calculated using Eq. (1) [28].

$$X_{i} = 2[2X - (X_{max} + X_{min})] / (X_{max} - X_{min})$$
(1)

where  $X_i$  is the required coded value of a parameter X. X is any value of the parameter from  $X_{min}$  to  $X_{max}$ , where  $X_{min}$  is lower limit of the parameter and  $X_{max}$  is upper limit of the parameter.

The welded joints were sliced at the midsection to prepare pitting corrosions' test specimens. For pitting corrosion test, specimens of  $50 \times 50$  mm (width and length) were prepared to ensure exposure of 12 mm diameter circular area in the weld region to the electrolyte. The rest of the area was covered with an acid resistant lacquer. The specimen size and dimensions are given in Fig. 1. The specimen was polished strictly following surface metallographic procedures. The polarisation studies of the welds were carried out in 1 N HCl solution. Analar grade chemicals and double distilled water were used for preparation of the electrolyte. The schematic circuit diagram of the potentiodynamic polarization set up is shown in Fig. 2. A potentiostat (Make: AUTOLAB /PGSTAT12) was used for this study in conjunction with an ASTM standard cell and personal computer. The experiments were performed in 2 h duration, each at the scan rate of 1 millivolts/mm. The pitting corrosion rate was calculated by polarizing the specimen anodically and cathodically and bv extrapolating the Tafel regions of anodic and cathodic curves to the corrosion potential. The experimentally evaluated results are presented in Table 4.

Table 1. Chemical composition of AISI 304L (weight %).

С	Si	Mn	Р	S	Cr	Ni	Мо	Ti	Ν
0.021	0.35	1.27	0.030	0.001	18.10	8.02			0.053

Table 2 .Welding conditions.				
Power source	Secheron Micro Plasma Arc Welding Machine (Model:			
	PLASMAFIX 50E)			
Polarity	DCEN			
Mode of operation	Pulse mode			
Electrode	2% thoriated tungsten electrode			
Electrode Diameter	1mm			
Plasma gas	Argon & Hydrogen			
Plasma gas flow rate	6 Lpm			
Shielding gas	Argon			
Shielding gas flow rate	0.4 Lpm			
Purging gas	Argon			
Purging gas flow rate	0.4 Lpm			
Copper Nozzle diameter	1mm			
Nozzle to plate distance	1mm			
Welding speed	260mm/min			
Torch Position	Vertical			
Operation type	Automatic			

Table 3.	Important factors	and	their	levels.

					Levels		
Serial No.	Input Factor	Units	-2	-1	0	+1	+2
1	Peak Current	Amperes	6	6.5	7	7.5	8
2	Back Current	Amperes	3	3.5	4	4.5	5
3	Pulse rate	Pulses/Second	20	30	40	50	60
4	Pulse width	%	30	40	50	60	70



Fig. 1. Dimensions of corrosion test specimen.



Fig. 2. Block diagram for the experimental set up.

A- Reference Electrode C-Working Electrode (AISI 304L) E-Electrolyte (1 N Hcl) B- Auxiliary Electrode (Platinum) D- Auto lab/PGTAT12 F-D.E. Amplifier

Exp. No.	Peak Current	Back	Pulse rate	Pulse
	(Amperes)	current	(Pulses/	width
		(Amperes)	Second)	(%)
1	6.5	3.5	30	40
2	7.5	3.5	30	40
3	6.5	4.5	30	40
4	7.5	4.5	30	40
5	6.5	3.5	50	40
6	7.5	3.5	50	40
7	6.5	4.5	50	40
8	7.5	4.5	50	40
9	6.5	3.5	30	60
10	7.5	3.5	30	60
11	6.5	4.5	30	60
12	7.5	4.5	30	60
13	6.5	3.5	50	60
14	7.5	3.5	50	60
15	6.5	4.5	50	60
16	7.5	4.5	50	60
17	6.0	4.0	40	50
18	8.0	4.0	40	50
19	7.0	3.0	40	50
20	7.0	5.0	40	50
21	7.0	4.0	20	50
22	7.0	4.0	60	50
23	7.0	4.0	40	30
24	7.0	4.0	40	70
25	7.0	4.0	40	50
26	7.0	4.0	40	50
27	7.0	4.0	40	50
28	7.0	4.0	40	50
29	7.0	4.0	40	50
30	7.0	4.0	40	50
31	7.0	4.0	40	50

 Table 4. Typical design matrix.

Amperes, back current of 4.5 Amperes, pulse rate of 30 pulses/second and pulse width of 40 %.

### 3. Experimental results

The measured pitting corrosion rate for all the 31 samples as per typical design matrix is presented in Table 5.

According to the conducted experiments, it could be understood that the minimum pitting corrosion rate of 0.54929 mm/year was obtained for the peak current of 6.5. Figures 3(a) and 3(b) represent SEM images of weld fusion zone before and after pitting corrosion in 1N HCl. The white patches in the SEM image indicate the area subjected to pitting corrosion.

#### Table 5. Experimental results.

Experiment	Pitting	Corrosion
No.	corrosion rate	Rate
	(mm/Year)	(mm/Year)
	Experimental	Predicted
1	0.54120	0.66439
2	0.99950	1.01518
3	0.53390	0.54929
4	0.82320	0.90273
5	0.85370	0.90143
6	0.70170	0.74111
7	0.79770	0.79328
8	0.60120	0.63561
9	0.80370	0.81858
10	0.99020	1.06891
11	0.54110	0.57598
12	0.82740	0.82896
13	1.12450	1.11926
14	0.82460	0.85849
15	0.85000	0.88361
16	0.67440	0.62549
17	0.78280	0.71458
18	0.86260	0.80725
19	1.23460	1.12227
20	0.78540	0.77417
21	0.89920	0.77908
22	0.81610	0.81265
23	0.78220	0.66853
24	0.82250	0.81260
25	0.66290	0.64569
26	0.64746	0.64569
27	0.72800	0.64569
28	0.71500	0.64569
29	0.52060	0.64569
30	0.62900	0.64569
31	0.61690	0.64569

### 3.1. Developing mathematical model

The output response of the weld joint (Y) is a function of peak current (A), back current (B), pulse rate (C) and pulse width (D). It can be expressed as Eq. (2) [29-31].

$$Y = f(A, B, C, D)$$
(2)



Fig. 3(a). SEM image before corrosion.



Fig. 3(b). SEM image after corrosion.

The second order polynomial equation used to represent the response surface 'Y' is given in Eq. (3) [16]:

$$Y = b_0 + \sum b_i x_i + \sum b_{ii} x_i^2 + \sum b_{ii} x_i x_i + \epsilon$$
(3)

where  $b_i x_i$  indiacte linear terms,  $b_{ij}x_ix_j$  indicate interaction terms and  $b_{ii}x_i^2$  represent pure second order or quadratic effects.

Using MINITAB 14 statistical software package, the significant coefficients were determined and final model was developed using significant coefficients to estimate pitting corrosion rate values of weld joint.

The final mathematical model for pitting corrosion rate is given in Eq. (4).

Pitting corrosion rate (CR)

 $CR = 0.645694 + 0.023167X_1 -$ 

where  $X_1$ ,  $X_2$ ,  $X_3$  and  $X_4$  are the coded values of peak current, back current, pulse rate and pulse width.

# 3.2. Checking adequacy of the developed model

Adequacy of the developed model was tested using Analysis of Variance (ANOVA) test. In this technique, if the calculated value of the  $F_{ratio}$  of the developed model is less than the standard  $F_{ratio}$  (from F-table) value at a desired level of confidence (say 99%), then the model is said to be adequate within the confidence level. ANOVA test results are presented in Table 6 for all the models. According to the table, the developed mathematical models were found to be adequate at 99% confidence level. The value of coefficient of determination '  $R^2$ ' for the above developed models was found to be about 0.86.

	Pitting corrosion rate								
Source	DF	Seq SS	Adj SS	Adj MS	F	Р			
Regression	14	0.72098	0.72098	0.051499	7.11	0.000			
Linear	4	0.22746	0.22746	0.056866	7.85	0.001			
Square	4	0.20184	0.20184	0.050461	6.97	0.002			
Interaction	6	0.29168	0.29168	0.048613	6.71	0.001			
Residual Error	16	0.11590	0.11590	0.007244					
Lack-of-Fit	10	0.08727	0.08727	0.008727	1.83	0.237			
Pure Error	6	0.02863	0.02863	0.004772					
Total	30	0.83689							

 Table 6. ANOVA table for pitting corrosion rate.

where DF= Degrees of Freedom, SS=Sum of Squares, MS=Mean Square, F=Fishers ratio.

Figure 4 indicates scatter plots for pitting corrosion rate of the weld joint and reveals that

the actual and predicted values are close to each other within the specified limits.

# *3.3. Effect of welding parameters on pitting corrosion rate*

# 3.3.1. Main effects

The above developed mathematical model can be employed to predict the weld pitting corrosion rates and their relationship for the range of parameters used in the investigation by substituting their respective values in the coded form. Based on these models, effects of the process parameters on the weld pitting corrosion rates were computed and plotted, as depicted in Fig. 5.



Fig. 4. Scatter plot for pitting corrosion rate.

Figure 5 shows that the pitting corrosion rate decreased from 6 Amperes of peak current to

6.5 Amperes and thereafter it increased up to 8 Amperes. Pitting corrosion rates decreased from 3 Amperes of back current to 4.5 Amperes and then increased up to 5 Amperes. Moreover, pitting corrosion rates decreased from 20 pulses/second of pulse rate to 30 pulse/ second and there was increase of up to 60 pulses/second thereafter. They also decreased from 30 % of pulse width to 40% and then increased up to 70 %.

# 3.3.2. Interaction effects

The simultaneous effect of two parameters at a time on the output response is generally studied using contour plots and surface plots.

# 3.3.2.1. Contour plots

Contour plots play a very important role in studying the response surface. By generating contour plots using statistical software (MINITAB14) for response surface analysis, the most influencing parameter can be identified based on the orientation of contour lines. If the counter patterning of circular shaped counters occurs, it suggests the equal influence of both factors while elliptical contours indicate interaction of the factors.

Figs. 6(a) to 6(c) represent contour plots for pitting corrosion rates. From these plots, the interaction effect between the input process parameters and output response can be observed as:



Fig. 5. Main effects for pitting corrosion rate.

(i) Pitting corrosion rate was more sensitive to change in peak current than in the base current (Fig. 6(a)) since the contour lines were more diverted towards the peak current.

(ii) Pitting corrosion rate was sensitive to both pulse rate and peak current (Fig. 6(b)) since the contour lines were circular in shape.

(iii) Pitting corrosion rate was more sensitive to peak current than pulse width (Fig. 6(c)) since the contour lines were more diverted towards the peak current.

From the contour plots, it is clear that the peak current had more effect on corrosion rate.

#### 3.3.2.2. Surface plots

Surface plots help in locating maximum and minimum values of the response. The maximum value of the response is represented by the apex of the surface plot whereas the minimum value is indicated by nadir of the surface plot. Response surface plots clearly indicate the optimal response point. The optimum pitting corrosion rate of pulsed current MPAW welded AISI 304L was exhibited by the nadir of the response surface, as shown in Figs.7(a) to 7(c).

Figure7(a) shows the three dimensional response surface plot for pitting corrosion rate obtained from the regression model, assuming a pulse rate of 40 pulses/second and pulse width

of 50%. The minimum pitting corrosion rate was exhibited by the nadir of the response surface. It can be seen from the twisted plane of surface plot that the model contained an interaction. From the response plot, it could be identified that, at the peak current of 7 Amperes and base current of 4 Amperes, pitting corrosion rate was minimum.

Figure 7(b) depicts the three dimensional response surface plot for the response pitting corrosion rate obtained from the regression model, assuming base current of 4 Amperes and pulse width of 50 %. According to the response plot, it can be identified that, atpeak current of 6 Amperes and pulse rate of 20 pulse/ second, pitting corrosion rate was minimum.

Figure7(c) shows the three dimensional response surface plot for the response pitting corrosion rate obtained from the regression model, assuming base current of 4 Amperes and pulse rate of 40 pulse/second. It can be seen from the twisted plane of surface plot that the model contained an interaction. According to the response plot also, at peak current of 6.5 Amperes and pulse width of 40 %, pitting corrosion rate was minimum.

Based on the surface plots, at peak current of 6 Amperes, Back Current of 4 Amperes, Pulse rate of 40 pulse/second and pulse width of 50%, optimum pitting corrosion rate was obtained.



**Fig. 6(a).** Contour plot for corrosion rate (Peak current vs. Back current).

**Fig. 6(b).** Contour plot for corrosion rate (Peak current vs. Pulse rate).



**Fig. 6(c).** Contour plot for corrosion rate (Peak current vs. Pulse width).



**Fig. 7(b).** Surface plot for corrosion rate (Peak current vs. Pulse rate).

#### 4. Conclusions

A five level, four factor full factorial design matrix based on the central composite rotatable design technique was used for the development of mathematical models to predict the pitting corrosion rate of AISI 304L Austenitic stainless sheets welded by pulsed current micro plasma arc welding process. From the contour plots, it was observed that peak current was the most dominating parameter which affected pitting corrosion rate compared to other parameters. According to the surface plots, minimum obtained pitting corrosion rate was 0.64569 mm/Year for the input parameter combination **Fig. 7(c).** Surface plot for corrosion rate (Peak current vs. pulse width).

Fig. 7(a). Surface plot for corrosion rate

(Peak current vs. Back current).

of peak current of 7Amperes, back current of 4 Amperes, pulse rate of 40 pulses /second and pulse width of 50% whereas the experimental value obtained for the above input parameter combination was 0.52060 mm/Year. It is very clear that the experimental and predicated values were close to each other.

#### 5. Acknowledgements

The authors would like to thank Shri. R. Gopla Krishnan, Director, M/s Metallic Bellows (I) Pvt Ltd, Chennai, India for his support during the preparation of this manuscript.

# References

- M. Balasubramanian, V. Jayabalan and V. Balasubramanian, "Effect of process parameters of pulsed current tungsten inert gas welding on weld pool geometry of titanium welds", *Acta Metall. Sin.(Engl. Lett.)*, Vol. 23, No. 4, pp. 312-320, (2010).
- B. Balasubramanian, V. Jayabalan and V. Balasubramanian, "Optimizing the Pulsed Current Gas Tungsten Arc Welding Parameters", *J. Mater. Sci. Technol*, Vol. 22, No. 6, pp. 821-825, (2006).
- [3] K. Siva Prasad, Ch. Srinivasa Rao and D. Nageswara Rao, "Prediction of Weld Pool Geometry in Pulsed Current Micro Plasma Arc Welding of SS304L Stainless Steel Sheets", *International Transaction Journal of Engineering, Management & Applied Sciences & Technologies*, Vol. 2, No. 3, pp. 325-336, (2011).
- [4] K. S. Prasad, Ch. Srinivasa Rao and D. Nageswara Rao, "A Study on Weld Quality Characteristics of Pulsed Current Micro Plasma Arc Welding of SS304L Sheets", *International Transaction Journal of Engineering, Management & Applied Sciences & Technologies*, Vol. 2, No. 4, pp. 437-446, (2011).
- [5] K. S. Prasad, Ch. Srinivasa Rao and D. Nageswara Rao, "Optimizing Pulsed Current Micro Plasma Arc Welding Parameters to Maximize Ultimate Tensile Strength of SS304L Sheets Using Hooke and Jeeves Algorithm", Journal of Manufacturing Science & Production (DyGuter"), Vol. 11, No. 1-3, pp. 39-48, (2011).
- [6] K. S. Prasad, Ch. Srinivasa Rao and D. Nageswara Rao, "Effect of Process Parameters of Pulsed Current Micro Plasma Arc Welding on Weld Pool Geometry of AISI 304L Stainless Steel Sheets", *Journal of Materials & Metallurgical Engineering*, Vol. 2, No. 1, pp. 37-48, (2012).
- [7] K. S. Prasad, Ch. Srinivasa Rao and D. Nageswara Rao, "Establishing Empirical

Relations to Predict Grain Size and Hardness of Pulsed Current Micro Plasma Arc Welded SS 304L Sheets", *American Transactions on Engineering* & *Applied Sciences*, Vo. 1, No. 1, pp. 57-74, (2012).

- [8] K. S. Prasad, Ch. Srinivasa Rao and D. Nageswara Rao, "Effect of pulsed current micro plasma arc welding process parameters on fusion zone grain size and ultimate tensile strength of SSS304L sheets", *International Journal of Lean Thinking*, Vo. 3, No. 1, pp. 107-118, (2012).
- [9] N. A. Falleiros and S. Wolynec, "Correlation between Corrosion Potential and Pitting Potential for AISI 304L Austenitic Stainless Steel in 3.5% NaCl Aqueous Solution", *Materials Research*, Vol. 5, No. 1, pp. 77-84, (2002).
- [10] B. Tsaneva, L. Fachikov, Y. Marcheva, M. Lukajcheva and B. Kostadinov, "Corrosion of Chromium Manganese-Nitrogen Steels In Chloride Media", *Journal of the University of Chemical Technology and Metallurgy*, Vol. 42, No. 2, pp. 163-168, (2007).
- [11] F. Fauvet, R. Balbaud, Q. Robin, T. Tran, A. Mugnier and D. Espinoux, "Corrosion mechanisms of austenitic stainless steels in nitric media used in reprocessing plants", *Journal of NuclearMaterials*, Vol. 375, pp. 52-64, (2008).
- [12] D. J. Lee, K. H. Jung, J. H. Sung, Y. H. Kim, K. H. Lee, J. U. Park, Y. T. Shin and H. W. Lee, "Pitting corrosion behavior on crack property in AISI 304L weld metals with varying Cr/Ni equivalent ratio", *Materials and Design*, Vol. 30, No. 8, pp. 3269-3273, (2009).
- [13] A. S. Afolabi, "Effect of Electric Arc Welding Parameters on Corrosion Behavior of Austenitic Stainless Steel in Chloride Medium", AU J. T., Vol. 11, No. 3, pp. 171-180, (2009).
- [14] Y. Prawoto, K. Ibrahim and W. B. Wan Nik, "Effect of PH and chloride concentration on the corrosion of Duplex Stainless Steel", *The Arabian Journal for*

*Science and Engineering*, Vol. 34, No. 2C, pp. 115-127, (2009).

- [15] G. Suresh, U. Kamachi Mudali and Baldev Raj, "Corrosion monitoring of type 304L stainless steel in nuclear nearhigh level waste by electrochemical noise", J. Appl. Electrochem, Vol. 41, pp. 973-981, (2011).
- [16] Y. Ait Albrimi, A. Eddib, J. Douch, Y. Berghoute, M. Hamdani and R. M. Souto, "Electrochemical Behaviour Of Aisi 316 Austenitic Stainless Steel In Acidic Media Containing Chloride Ions", *Int. J. Electrochem. Sci.*, Vol. 6, pp. 4614-4627, (2011).
- [17] Md. Asaduzzaman, C. M. Mustafa and M. Islam, "Effects of Concentration of Sodium Chloride on the Pitting Of AISI-304L Corrosion Behavior Austenitic Stainless Steel", Chemical Industrv & Chemical Engineering Quarterly, Vol. 17, No. 4, pp. 477-483, (2011).
- [18] M. Saadawy, Kinetics of Pitting Dissolution of Austenitic Stainless Steel 304 in Sodium Chloride Solution, International Scholarly Research Network, "ISRN Corrosion", Vol. 2012, Article ID 916367, pp. 1-5, (2012).
- [19] K. Marimuthu and N. Murugan, "Prediction and optimization of weld bead geometry of plasma transferred arc hardfaced valve seat rings", *Surf Eng*, Vol. 19, No. 2, pp. 143-149, (2003).
- [20] V. Gunaraj and N. Murugan, "Prediction and comparison of the area of the heat affected zone for the bead-on-plate and bead-on joint in SAW of pipes", *J. Mat. Proc. Tech.*, Vol. 95, pp. 246-261, (1999).
- [21] V. Gunaraj and N. Murugan, "Application of response surface methodology for predicting weld quality in saw of pipes", J. Mat. Proc. Tech., Vol. 88, pp. 266-275, (1999).

- [22] S. C. Juang, Y. S. Tarng, "Process parameter selection for optimizing the weld pool geometry in the TIG welding of stainless steel", *J. Mat. Proc. Tech.*, Vol. 122, pp. 33-37, (2002).
- [23] T. T. Allen, R. W. Richardson, D. P. Tagliable and G. P. Maul, "Statistical process design for robotic GMA welding of sheet metal", *Weld. J.*, Vol. 81, No. 5, pp. 69-s-172-s, (2002).
- [24] I. S. Kim, J. S. San and Y. J. Jeung, "Control and optimization of bead width for mutipass-welding in robotic arc welding processes", *Aus. Weld. J.*, Vol. 46, pp. 43-46, (2001).
- [25] D. Kim, M. Kang, S. Rhee, "Determination of optimal welding conditions with a controlled random search procedure", *Weld J.*, pp. 125-s-130-s, (2005).
- [26] D. C. Montgomery, "Design and analysis of experiments", 3<sup>rd</sup> Edition, NewYork: John Wiley & Sons, pp. 291-295, (1991).
- [27] G. E. P. Box, W. H. Hunter and J. S. Hunter, "Statistics for experiments", New York: John Wiley & Sons, pp. 112-115, (1978).
- [28] S. Babu, T. SenthilKumar and V. Balasubram anian, Optimizing pulsed current gas tungsten arc welding parameters of AA6061 aluminium alloy using Hooke and Jeeves algorithm, "Trans .Nonferrous Met. Soc."China, Vol. 18, pp. 1028-1036, (2008).
- [29] W. G. Cochran and G. M. Cox, "Experimental Designs", London: John Wiley & Sons Inc, (1957).
- [30] T. B. Barker, "*Quality by experimental design*", Marcel Dekker: ASQC Quality Press, (1985).
- [31] W. P. Gardiner and G. Gettinby, "Experime ntal design techniques in statistical practice", Chichester: Horwood, (1998).